

Supporting Information

The Ideal Gas Thermochemistry of Oxonium Cations

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Table S1 7 term NASA polynomials of oxonium compounds (thermal electron convention) of protonated compounds and radical cations.

CH3O+ Methylidene Oxonium cation	CHOH2+	HF298=1042.05±8. kJ/mol	Burcat G3B3	
CHOH2+ Oxonium	T 5/15C 1.H 3.O 1.E -1.G	298.150 6000.000 1000.	1	
	4.39380166E+00 7.16728474E-03-2.46127134E-06 3.85124272E-10-2.25476964E-14		2	
	1.23503552E+05 3.25046236E-01 1.69308260E+00 1.45930724E-02-9.09655507E-06		3	
	1.76040858E-09 5.12383125E-13 1.24252331E+05 1.42806460E+01 1.25329093E+05		4	
CH4O*+ Methylene Oxonium cation	CH2OH2*+	HF298=825.9±4.5 kJ/mol	Ruscic ATcT 2013	
CH2OH2+ Oxonium	T 5/15C 1.H 4.O 1.E -1.G	298.150 6000.000 1000.	1	
	5.04696028E+00 8.83268034E-03-2.96312821E-06 4.55893988E-10-2.63627238E-14		2	
	9.72441646E+04-1.99964848E+00 1.59132994E+00 2.00143834E-02-1.60986921E-05		3	
	6.60975763E-09-7.70735214E-13 9.80978732E+04 1.53918843E+01 9.93323720E+04		4	
CH5O+ Methyl Oxonium (Rydberg cation)	CH3OH2+	HF298=583.7±8. kJ	Burcat G3B3	
CH5O+ CH3OH2+	T11/14C 1.H 5.O 1.E -1.G	298.150 6000.000 1000.	1	
	4.12018251E+00 1.23179143E-02-4.21315216E-06 6.57462537E-10-3.84191498E-14		2	
	6.81422088E+04 1.71193370E+00 2.63252920E+00 9.23812483E-03 1.43800480E-05		3	
	-2.22087922E-08 8.80207817E-12 6.89204876E+04 1.11834286E+01 7.02032952E+04		4	
C2H5O+ Methyl Methylene Oxonium	CH3-O=CH2+	HF298=676.9±8. kJ	Burcat G3B3	
C2H5O+ CH3-O=CH2+	T11/14C 2.H 5.O 1.E -1.G	298.150 6000.000 1000.	1	
	5.23682452E+00 1.45674929E-02-5.15192045E-06 8.22867349E-10-4.88896791E-14		2	
	7.87093166E+04-3.31148010E+00 3.32715925E+00 7.43454980E-03 2.71006098E-05		3	
	-3.59610292E-08 1.35250850E-11 7.99157892E+04 9.75450337E+00 8.14129756E+04		4	
C2H5O+ Protonated Oxirane [CH2-CH2]OH+	HF298=715.07±0.62 kJ	Ruscic ATcT D 2013		
C2H5O+ (CH2CH2)OH+T12/14C	2.H 5.O 1.E -1.G	298.150 6000.000 1000.	1	
	5.70791482E+00 1.39321006E-02-4.87755699E-06 7.73937429E-10-4.57782079E-14		2	
	8.30274882E+04-8.39211861E+00-5.22964914E-01 2.30971327E-02 5.20358577E-06		3	
	-2.33955821E-08 1.11084462E-11 8.51270059E+04 2.57116207E+01 8.60026628E+04		4	
C2H6O*+ Methyl Methylene Proton Oxonium	CH3-OH-CH2*+	HF298=797.6±8. kJ	Burcat	
C2H6O+ CH3-OH-CH2+T 5/15C	2.H 6.O 1.E -1.G	298.150 6000.000 1000.	1	
	6.85380737E+00 1.52932218E-02-5.29722518E-06 8.34043510E-10-4.90531913E-14		2	
	9.27293683E+04-1.05689378E+01 2.50821748E+00 2.11152697E-02 2.81122644E-06		3	
	-1.66091478E-08 7.80922851E-12 9.42502585E+04 1.34228867E+01 9.59322919E+04		4	
C2H7O+ DiMethyl Oxonium (CH3)2-OH+	HF298=562.24±8. kJ	Burcat G3B3		
C2H7O+ (CH3)2OH+T 11/14C	2.H 7.O 1.E -1.G	298.150 6000.000 1000.	1	
	5.81266575E+00 1.89420633E-02-6.61986229E-06 1.04878304E-09-6.19569490E-14		2	
	6.44582085E+04-7.03041901E+00 3.49195920E+00 9.77638270E-03 3.50079732E-05		3	
	-4.70880441E-08 1.79334219E-11 6.59209412E+04 8.89898119E+00 6.76213030E+04		4	
C2H7O+ Ethyl Oxonium C2H5OH2+	HF298=527.76±8. kJ	Burcat G3B3		
C2H7O+ C2H5OH2+	T11/14C 2.H 7.O 1.E -1.G	298.150 6000.000 1000.	1	

6.18891803E+00	1.79926340E-02	-6.24009542E-06	9.83369694E-10	-5.78725674E-14	2
6.03219647E+04	-7.85592076E+00	2.86917130E+00	1.73908141E-02	1.49889555E-05	3
-2.77297126E-08	1.13477254E-11	6.17634220E+04	1.17829996E+01	6.34748169E+04	4

C3H7O+ PropyleneOxide Oxonium CH₃[CH-CH₂]OH+ HF298=648.0±8. kJ Burcat G3B3

C3H7O+ MeOxoProt	T12/14C	3.H	7.O	1.E -1.G	298.150	6000.000	1000.	1
8.00791650E+00	1.98766698E-02	-7.00852891E-06	1.11739544E-09	-6.63134336E-14	2			
7.37876008E+04	-1.91729839E+01	1.71414238E-01	3.05635474E-02	6.92713138E-06	3			
-2.93671651E-08	1.36322114E-11	7.65168794E+04	2.40418371E+01	7.79360419E+04	4			

C3H7O+ CycloTriMethyleneOxide Oxonium [CH₂CH₂CH₂]OH+ HF298=640.85±8. kJ Burcat G3B3

C3H7O+ CyOxonium	T12/14C	3.H	7.O	1.E -1.G	298.150	6000.000	1000.	1
7.43398460E+00	2.04359385E-02	-7.22191223E-06	1.15327670E-09	-6.85237610E-14	2			
7.29613893E+04	-1.78357142E+01	-1.51555883E+00	3.27175286E-02	8.98977573E-06	3			
-3.44443374E-08	1.59853557E-11	7.60548693E+04	3.14505091E+01	7.70760995E+04	4			

C3H8O*+ Dimethyl Methylene Oxonium (CH₃)₂O-CH₂*+ HF298=772.8±8. kJ Burcat G3B3

C3H8O+ (CH3)2O=CH2T10/14C	3.H	8.O	1.E -1.G	298.150	6000.000	1000.	1
8.67839202E+00	2.17590005E-02	-7.63698179E-06	1.21342751E-09	-7.18282028E-14	2		
8.86610510E+04	-1.98461405E+01	3.64462504E+00	2.27457201E-02	1.85409505E-05	3		
-3.57734005E-08	1.47841149E-11	9.07509991E+04	9.46516110E+00	9.29487126E+04	4		

C3H9O+ TriMethyl Oxonium (CH₃)₃O+ HF298=541.716±8. kJ/mol Burcat G3B3

C3H9O+ (CH3)3O+ T10/14C	3.H	9.O	1.E -1.G	298.150	6000.000	1000.	1
7.59728406E+00	2.54784347E-02	-8.99285110E-06	1.43435890E-09	-8.51363342E-14	2		
6.09096282E+04	-1.64610245E+01	4.51970904E+00	1.18541575E-02	4.97969882E-05	3		
-6.52816266E-08	2.45428621E-11	6.29561255E+04	5.10913040E+00	6.51530878E+04	4		

C3H9O+ n-Propanol Protonated Oxonium n-C3H7OH2+ HF298=501.85±8. kJ Burcat G3B3

C3H9O+ nC3H7OH2+ T 5/15C	3.H	9.O	1.E -1.G	298.150	6000.000	1000.	1
9.44845876E+00	2.35494635E-02	-8.48735271E-06	1.36974058E-09	-8.18959901E-14	2		
5.54563517E+04	-2.49192675E+01	1.57568968E+00	2.98650264E-02	1.75864292E-05	3		
-4.05058476E-08	1.72771164E-11	5.84771733E+04	1.97302170E+01	6.03578563E+04	4		

C3H9O+ i-Propanol Protonated Oxonium i-C3H7OH2+ HF298=481.8±8. kJ Burcat G3B3

C3H9O+ iC3H7OH2+ T 5/15C	3.H	9.O	1.E -1.G	298.150	6000.000	1000.	1
9.57409469E+00	2.31713068E-02	-8.17316606E-06	1.29960391E-09	-7.68969622E-14	2		
5.19914246E+04	-2.52943675E+01	2.25481526E+00	3.15666564E-02	9.08108465E-06	3		
-3.10362905E-08	1.39286765E-11	5.46444712E+04	1.55201868E+01	5.67452562E+04	4		

C4H9O+ TetraHydroFuran Oxonium [CH₂CH₂CH₂CH₂]OH+ HF298=529.07±8. kJ Burcat G3B3

C4H9O+ TetHydFuranT12/14C	4.H	9.O	1.E -1.G	298.150	6000.000	1000.	1
9.27420638E+00	2.68100376E-02	-9.51325112E-06	1.52337781E-09	-9.06884102E-14	2		
5.83840237E+04	-2.82047764E+01	-2.08366085E+00	4.20226784E-02	1.12339571E-05	3		
-4.29010524E-08	1.97317600E-11	6.23614523E+04	3.45188773E+01	6.36317714E+04	4		

C4H11O+ Ethyl DiMethyl Oxonium C₂H₅-O-(CH₃)₂+ HF298=498.±8. kJ Burcat G3B3

C4H11O+ Oxonium T10/14C	4.H	11.O	1.E -1.G	298.150	6000.000	1000.	1
9.41866464E+00	3.13815691E-02	-1.11039894E-05	1.77410334E-09	-1.05428724E-13	2		
5.46000657E+04	-2.53557791E+01	4.76501693E+00	1.78290766E-02	5.63814406E-05	3		
-7.68147655E-08	2.92041503E-11	5.73201177E+04	5.46915280E+00	5.98933671E+04	4		

C4H11O+ DiEthyl Oxonium (C₂H₅)₂OH+ HF298=468.08±8. kJ/mol Burcat G3B3

C4H11O+ (C2H5)2OH+T11/14C	4.H	11.O	1.E -1.G	298.150	6000.000	1000.	1
9.93034521E+00	3.02853390E-02	-1.06660779E-05	1.69877827E-09	-1.00731034E-13	2		
5.09606767E+04	-2.83500513E+01	3.23373030E+00	3.04473021E-02	2.67898693E-05	3		
-4.94050560E-08	2.00818078E-11	5.38304451E+04	1.10075657E+01	5.62964023E+04	4		

C5H11O+ PentaHydroxyPyran Oxonium [CH₂CH₂CH₂CH₂CH₂]OH+ HF298=510.96±8. kJ Burcat G3B3

C5H11O+ PenHydPyrat 1/15C	5.H	11.O	1.E -1.G	298.150	6000.000	1000.	1
1.10492507E+01	3.32787293E-02	-1.18463934E-05	1.90101634E-09	-1.13335364E-13	2		

5.51041284E+04-3.79392351E+01-2.04330480E+00	4.83518529E-02	1.87019607E-05	3
-5.53878992E-08	2.46466580E-11	5.98475677E+04	4
3.50599406E+01	6.14548542E+04		

C5H13O+ Methyl DiEthyl Oxonium $\text{CH}_3\text{-O-(C}_2\text{H}_5)_2^+$ HF298=455.7±8. kJ Burcat G3B3

C5H13O+ CH3O(C2H5)T11/14C	5.H	13.O	1.E	-1.G	298.150	6000.000	1000.	1
1.23036206E+01	3.57854312E-02	-1.26513322E-05	2.02012270E-09	-1.19998347E-13				2
4.82194810E+04	-4.14071543E+01	2.26691809E+00	4.31421706E-02	2.28259869E-05				3
-5.39722310E-08	2.30749993E-11	5.21111296E+04	1.56938999E+01	5.48104458E+04				4

C6H15O+ TriEthyl Oxonium $(\text{C}_2\text{H}_5)_3\text{O}^+$ HF298=414.7±8. kJ/mol Burcat G3B3

C6H15O+ (C2H5)3O+ T11/14C	6.H	15.O	1.E	-1.G	298.150	6000.000	1000.	1
1.38748821E+01	4.23805960E-02	-1.50162400E-05	2.40134262E-09	-1.42793078E-13				2
4.23572955E+04	-5.00923764E+01	3.65827166E+00	4.63430467E-02	2.98234566E-05				3
-6.17841058E-08	2.54571186E-11	4.65724949E+04	9.09434099E+00	4.98764209E+04				4

* Oxoniums that are radical cations or distonic ions.